

riag Oberflächentechnik AG · Postfach 169 · CH-9545 Wängi TG

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riag PMn 922

Acidic phosphating agent generating manganese phosphate coatings in immersion processes

The riag PMn 922 is a phosphating process which produces uniform coatings of manganese phosphate on steel and ferrous materials.

Properties

- Easy to operate, broad working area
- Excellent corrosion protection
- Excellent absorption of anticorrosion agents
- Produces uniform micro- or coarse crystalline coatings
- Coating weight of up to 40 g/m² possible

Make-up

	Range	Optimum
riag PMn 922 Additive	80 – 140 mL/L	100 mL/L
Total acid (TA)	12 – 13 points	12.5 pts
Free acid (FA)	1.5 – 2.5 points	2.0 pts
Ratio	6 – 9	7.5
Iron (II)-content	0.1 – 0.4 %	0.2 %
Temperature	93 – 100 °C	as required
Dip time	15 – 30 min.	as required

Range

Fill tank with water up to ¾ of the final volume. Add the required amount of riag PMn 922 Additive and fill up with water to the final volume. Heat the agent up to a working temperature of 70 – 75 °C. By adding steel wool (120 g/100 L bath) or steel scraps (600 g/100 L bath) the iron content is increased to a minimum amount. This takes about 30 – 60 minutes. The ratio of total acid to free acid is being determined by analysis and adjusted to the optimum. After heating up to the minimal temperature the electrolyte is ready for use.

Operating parameters

Temperature $93 \,^{\circ}\text{C} \, (93 - 100 \,^{\circ}\text{C})$

Make-up Typically 100 mL/L riag PMn 922 Additive

Total acid (TA) 12.5 points (12 – 13 points)

Free acid (FA) 2.0 points (1.5 - 2.5 points)

Ratio 7.5 (6 - 9) The ratio is determined by the amount of total acid compared

to free acid

Iron content 0.2% (0.1 – 0.4%) The correct amount of iron is very important in order

to get satisfying coatings. During longer shutdown times the iron content

has to be checked before start-up.

pH - Value 2.0 – 2.4 Phosphating processes are usually controlled by TA, FA

and / or the ratio of TA to FA

Agitation Not essential but beneficial (involves continues sludge removal)

Tank Stainless steel or lined steel tanks. The bottom of the tank should either

have a slope, a sludge channel or a disc valve.

Heating Stainless steel heaters or PTFE lined. Optimal heating results can be

achieved by heating alongside the tank without locally overheating.

Thermostatic temperature control is necessary.

Cooling Not necessary

Exhaustion Absolutely necessary

Replenishment

To increase the total acid by 1 point add 9.5 mL/L **riag PMn 922 Additive**. This will increase the free acid by 0.35 points. Should the content of free acid get too high adjust it by adding small amounts of sodium hydroxide (Caution!). Ideally replenishment is done with a dosing pump.

Procedure

- Cleaning and degreasing
- 2. Rinsing
- 3. Pickling (only necessary if parts are very rusty)
- 4. Rinsing / Activation in riag PMn 920
- 5. Phosphating in riag PMn 922
- 6. Rinsing
- 7. Neutralising
- 8. Anti-corrosive oil and / or drying

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Desludging

Phosphating processes produce sludge which has to be removed regularly. After desludging or a partial electrolyte loss water has to be added. The electrolyte has to be analysed and replenished accordingly.

Corrosion protection

Ask for our non-committal advice

Waste water treatment / Environmental protection

All concentrates, rinse waters and waste solution must be treated and discharged in accordance with local effluent control regulations. The product contains acids; please consider your own safety. Information can be gleaned from the material safety data sheets.

Safety instructions

For further information please consult the material safety data sheets. Chemicals must not be stored below 10°C.

Liability

This instruction manual was compiled with reference to the state of the art and all current standards, and is based on the long-term knowledge and experience of riag. However, riag cannot monitor compliance with this instruction manual and the methods described herein at the customer/end-user's premises. Work carried out with riag products must be adapted accordingly to meet local conditions. In particular, riag cannot accept liability for damage, loss or cost incurred due to a failure to adhere to this instruction manual, improper application of the methods, unauthorised technical modifications, insufficient maintenance or the absence of maintenance in respect of the requisite technical hardware or equipment, or in the event of use by unqualified personnel. riag is not liable for damage or loss caused by riag or its employees except where intention or gross negligence can be proved. riag furthermore reserves the right to make changes in relation to products, methods and the instruction manual without prior notice.

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This transaction is governed by material Swiss law (Law of Obligations), excluding private international law (conflict of laws) and intergovernmental treaties, specifically the CISG.

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Analysis (Analytical methods)

Sample preparation:

The sample must be taken from a well-mixed location and allowed to cool down to 25 °C. If needed the sludge has to be removed by decantation.

Free acid in riag PMn 922

Reagents: Sodium hydroxide solution 0.1 mol/L

Bromophenol blue (1 % in ethanol)

Procedure: 10 mL phosphate solution are transferred via pipette into a

250 mL beaker, add

50 mL deion. water, add

5 drops bromophenol solution

Titrate with sodium hydroxide solution from red to blue.

The colour change is rather dragging.

Calculation: Use of NaOH in mL : 5 = free acid (FA points)

Total acid in riag PMn 922

Reagents: Sodium hydroxide solution 0.1 mol/L

Phenolphthalein (1 % in ethanol)

Procedure: 2 mL phosphate solution are transferred via pipette into a

250 mL beaker, add

10 mL deion. water, add

3 drops phenolphthalein solution

Titrate with sodium hydroxide solution from colourless to

pink.

Calculation: Use of NaOH in mL = total acid (TA points)

Use of NaOH in mL x 9.6 = mL/L riag PMn 922 Ratio = total acid (points) : free acid (points)

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Iron (II) in riag PMn 922

Reagents: Potassium permanganate 0.02 mol/L

Sulfuric acid (1:1)

Procedure: 10 mL phosphate solution are transferred via pipette into a

> 250 mL beaker, add

50 mL deion. water, add

1 mL sulfuric acid

Titrate with potassium permanganate from colourless to purple. Purple colour has to stay for 15 seconds.

Calculation: Use of potassium permanganate in mL x 0.05 = % Iron (II)

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